

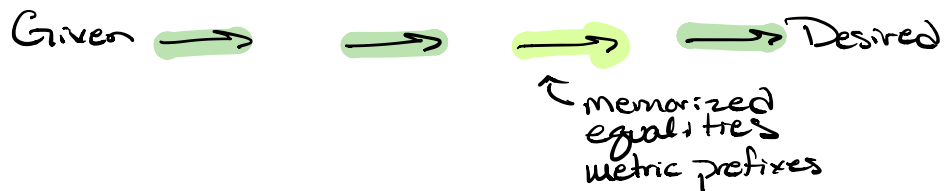
# Review for Exam 1

## - Conversions (Dimensional Analysis problems)

### Process

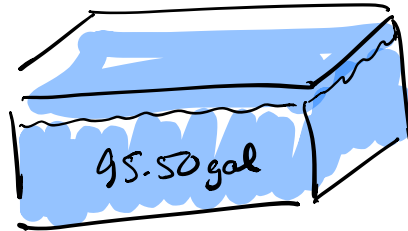
- parse the problem - word problem
  - \* identify the **given**
  - \* identify the **desired**
  - \* identify any **equalities** (conversion factors)

• Build a road map

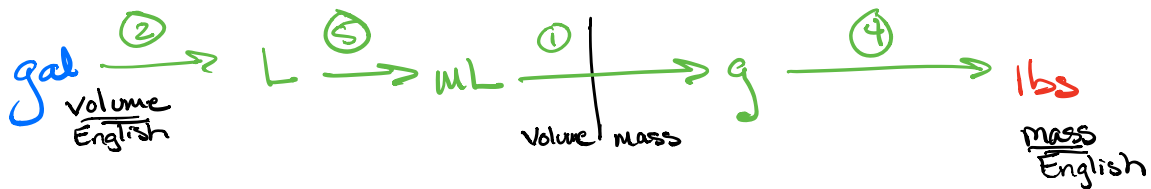


- Write out the calculation with equalities as conversion factors
- Calculate answer
- Apply Sig Figs Rules
- Check units
- Write answer w/ proper units

A fish tank hold <sup>not needed</sup> 100 gallons of water. Salt water has a density of <sup>Equality</sup> 1.035 g/mL. <sup>Desired</sup> How many pounds of water will the fish tank weigh if filled to <sup>Given</sup> 95.50 gallons?



⇒ Conversion factors usually have 2 or more units (g/mL)



### Equalities

①  $1.035 \text{ g} = 1 \text{ mL}$  Density  
mass SI      volume SI

3 key Equalities everyone should memorize

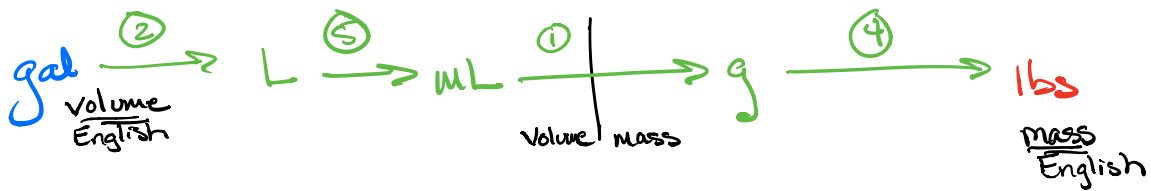
- ②  ~~$1 \text{ gal} = 3.785 \text{ L}$  volume~~
- ③  ~~$1 \text{ in} = 2.54 \text{ cm}$  length x~~
- ④  ~~$1 \text{ lb} = 453.6 \text{ g}$  mass~~

### SI Conversions

⑤  ~~$1 \text{ L} = 1000 \text{ mL}$~~

$$1.035 \text{ g} = 1 \text{ mL}$$

$$1 \text{ mL} = 1.035 \text{ g}$$



$$95.50 \text{ gal} \times \frac{3.785 \text{ L}}{1 \text{ gal}} \times \frac{1000 \text{ mL}}{1 \text{ L}} \times \frac{1.035 \text{ g}}{1 \text{ mL}} \times \frac{1 \text{ lbs}}{453.6 \text{ g}} = \text{lbs}$$

$$95.50 \times 3.785 \times 1000 \times 1.035 \div 453.6 = 824.77703373 \text{ lbs}$$

$$= \boxed{824.8 \text{ lbs}}$$

$$\frac{8.63 \text{ cm} \times 92.7 \text{ cm} \times 6.06 \text{ cm}}{72.6 \text{ s} \times 301.7 \text{ m}} =$$

$$\frac{A \times B \times C}{D \times E} = \frac{(A \times B \times C)}{(D \times E)} = (A \times B \times C) \times \frac{1}{(D \times E)}$$

$$= \boxed{(A \times B \times C) \div (D \times E)}$$

$$\frac{A \times B \times C}{D \times E} = \frac{A}{1} \times \frac{B}{1} \times \frac{C}{1} \times \frac{1}{D} \times \frac{1}{E}$$

$$= A \times B \times C \div \frac{D}{1} \div \frac{E}{1}$$

$$= \boxed{A \times B \times C \div D \div E}$$

A certain medication has a dose of  $1.62 \mu\text{g}/\text{kg}\cdot\text{day}$  of body weight. If a patient weighs  $165 \text{ lbs}$ , how many  $\mu\text{g}$  of medication should be prescribed?

Given

$$M = \text{Mega} \times 10^6$$

$$k = \text{kilo} \times 10^3$$

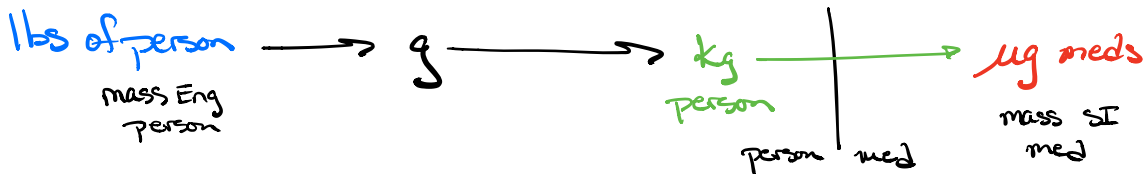
---


$$c = \text{centi} \times 10^{-2}$$

$$m = \text{milli} \times 10^{-3}$$

$$\mu = \text{micro} \times 10^{-6}$$

### Road Map



mass key  $1 \text{ lb} = 453.6 \text{ g}$  (4 SF)  
 $1 \text{ lb} = 2.2 \text{ kg}$  (2 SF)

$$1000 \text{ g} = 1 \text{ kg}$$

$$165 \text{ lbs} \times \frac{453.6 \text{ g}}{1 \text{ lbs}} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{1.62 \text{ } (\mu\text{g}) \text{ med}}{1 \text{ kg person}} =$$

$$165 \times 453.6 \div 1000 \times 1.62 = 121.24728 \text{ } \mu\text{g} \text{ medication}$$

or

$$165 \times 453.6 \times 1.62 \div 1000$$

$$= 121 \text{ } \mu\text{g} \text{ medication needed}$$

Exact

Definitions  
or  
Counted

Have infinite  
sig figs in the  
way we apply sig  
figs.

$$1 \text{ ft} = 12 \text{ in}$$

$$3 \text{ ft} = 1 \text{ yard}$$

$$1000 \text{ mm} = 1 \text{ m}$$

$$1000 \text{ L} = 1 \text{ kL}$$

units of measure within  
the same system

How the system is defined

⇒ Definition.

\*\* exception

$$1 \text{ in} = 2.54 \text{ cm} \text{ Definition}$$

Experiment in 1970

$$1 \text{ m} = \underbrace{2.5400000}_{\text{exact}} \times \text{cm} \Rightarrow \text{made it a definition}$$

Not Exact (Contain precision)

measured

Have sig figs

$$\text{Eng} \rightarrow \text{SI} \quad 4 \text{ SF}$$

$$1 \text{ lb} = 453.6 \text{ g} \quad 4 \text{ SF}$$

$$1 \text{ gal} = 3.785 \text{ L}$$

## SI System

$$T = \text{Tera} \times 10^{12}$$

$$G = \text{Giga} \times 10^9$$

$$M = \text{Mega} \times 10^6$$

$$k = \text{kilo} \times 10^3$$

$$c = \text{centi} \times 10^{-2}$$

$$m = \text{milli} \times 10^{-3}$$

$$\mu = \text{micro} \times 10^{-6}$$

$$n = \text{nano} \times 10^{-9}$$

$$\text{Base} \times 10^0 = \times 1$$

$$T = \text{Tera} \times 10^{12}$$

$$G = \text{Giga} \times 10^9$$

$$M = \text{Mega} \times 10^6$$

$$k = \text{kilo} \times 10^3$$

$$c = \text{centi} \times 10^{-2}$$

$$m = \text{milli} \times 10^{-3}$$

$$\mu = \text{micro} \times 10^{-6}$$

$$n = \text{nano} \times 10^{-9}$$

$$1 \text{ Mbase} = 1 \times 10^6 \text{ base} = 1,000,000 \text{ base}$$

$$1 \text{ kbase} = 1 \times 10^3 \text{ base} = 1000 \text{ base}$$

$$1 \text{ cbase} = 1 \times 10^{-2} \text{ base} = 0.01 \text{ base}$$

$$100 \text{ cbase} = 1 \text{ base}$$

$$1 \text{ mbase} = 1 \times 10^{-3} \text{ base} = 0.001 \text{ base}$$

$$1000 \text{ mbase} = 1 \text{ base}$$

Convert 36.2 mm into meters

$$36.2 \text{ mm} \times \frac{1 \times 10^{-3} \text{ m}}{1 \text{ mm}} = 36.2 \times \underset{EE}{(EE)^3} = \boxed{0.0362 \text{ m}}$$

$$36.2 \text{ mm} \times \frac{1 \text{ m}}{1000 \text{ mm}} = 36.2 \div 1000 = \boxed{0.0362 \text{ m}}$$

$$36.2 \text{ mm} \times \frac{0.001 \text{ m}}{1 \text{ mm}} = 36.2 \times 0.001 = \boxed{0.0362 \text{ m}}$$